

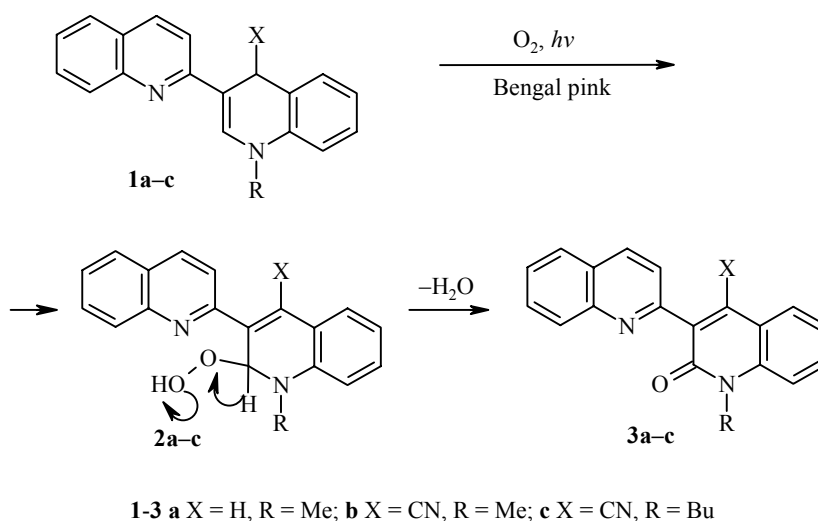
REACTION OF 1'-ALKYL- 1',4'-DIHYDRO-2,3'-BIQUINOLYLS WITH SINGLET OXYGEN

A. V. Aksenov

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In previous work [1, 2], we developed a series of methods for the synthesis of 1'-R-1',4'-dihydro-2,3'-biquinolyls **1**, which permitted us to study the properties of these compounds. In the present work, we report the reaction of **1** with singlet oxygen.

We have found that the oxidation of 1'-alkyl-1',4'-dihydro-2,3'-biquinolyls **1** by singlet oxygen at room temperature yields 1'-alkyl-1',2'-dihydro-2,3'-biquinolyl-2'-ones **3**. The mechanism of this reaction probably involves formation of hydroperoxide **2** in the first step with subsequent loss of water.



A stream of air was usually passed through a mixture of 2.5 mmol of **1** and 2 mg Bengal pink in 40 ml methanol in a quartz flask irradiated by a high-pressure lamp until the starting compound disappeared as indicated by thin-layer chromatography (0.5-1.5 h). The solvent was evaporated and the reaction product was purified by recrystallization.

1'-Methyl-1',2'-dihydro-2,3'-biquinolyl-2'-one (3a) was obtained in 77% yield; mp 174-175°C (ethanol) (174-175°C [3]). IR spectrum (neat), ν, cm^{-1} : 1602 (C=O). A mixed probe with an authentic sample gave an undepressed melting point.

Stavropol State University, 355009 Stavropol, Russia; e-mail: nauka@stavsru.ru. Translated from *Khimiya Geterotsiklicheskikh Soedinenii*, No. 10, pp. 1421-1422, October, 2001. Original article submitted March 19, 2001.

1'-Methyl-4'-cyano-1',2'-dihydro-2,3'-biquinolyl-2'-one (3b) was obtained in 81% yield; mp 221-222°C (benzene). IR spectrum (neat), ν , cm^{-1} : 1612 (C=O). ^1H NMR spectrum at 200 MHz (acetone- d_6), δ , ppm, J (Hz): 3.87 (3H, s, Me); 7.54 (1H, dd, $J_{56} = 8.23$, $J_{67} = 7.88$, 6-H); 7.72 (1H, dd, $J_{5'6'} = 7.91$; $J_{6'7'} = 7.88$; 6'-H); 7.77 (1H, d, $J_{7'8'} = 8.41$, 7-H); 8.00 (1H, d, $J_{34} = 8.54$, 3-H); 8.07 (1H, d, $J_{5'6'} = 7.91$, 5'-H); 8.11 (1H, d, $J_{56} = 8.23$, 5-H); 8.17 (1H, d, $J_{78} = 8.41$, 8-H); 8.47 (1H, d, $J_{34} = 8.54$, 4-H). Mass spectrum, m/z (70 eV): 311 [M^+] (100%). Found, %: C 77.34; H 4.16; N 13.45. $\text{C}_{20}\text{H}_{13}\text{N}_3\text{O}$. Calculated, %: C 77.16; H 4.21; N 13.50.

1'-Butyl-4'-cyano-1',2'-dihydro-2,3'-biquinolyl (3c) was obtained in 78% yield; mp 208-209°C (benzene). IR spectrum (neat), ν , cm^{-1} : 1618 cm^{-1} (C=O). ^1H NMR spectrum at 200 MHz (CDCl_3), δ , ppm, J (Hz): 1.18 (3H, t, $J = 7.68$, 1'- $\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$); 1.53 (2H, m, 1'- $\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$); 1.79, m, 1'- $\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$); 4.39 (2H, t, $J = 7.69$, 1'- $\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$); 7.42 (1H, dd, $J_{5'6'} = 7.91$, $J_{6'7'} = 7.88$, 7'-H); 7.48 (1H, d, $J_{7'8'} = 7.96$, 8'-H); 7.62 (1H, dd, $J_{56} = 7.88$, $J_{78} = 8.41$, 7-H); 7.90 (1H, d, $J_{5'6'} = 7.91$, 5'-H); 7.94 (1H, d, $J_{34} = 8.54$, 3-H); 8.19 (1H, d, $J_{56} = 8.23$, 5-H); 8.23 (1H, d, $J_{78} = 8.41$, 8-H); 8.30 (1H, d, $J_{34} = 8.54$, 4-H). Mass spectrum, m/z (70 eV): 353 [M^+] (88%). Found, %: C 78.34; H 5.32; N 11.69. $\text{C}_{23}\text{H}_{19}\text{N}_3\text{O}$: C 78.16; H 5.42; N 11.89.

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